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MEMORANDUM FOR PRS (In-House Publication)

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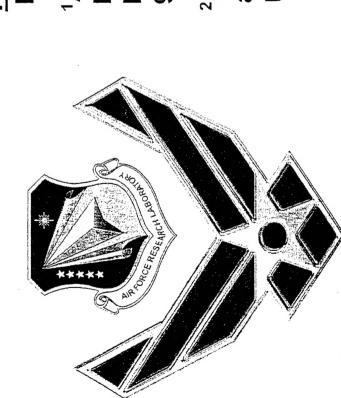
11 June 2001

SUBJECT: Authorization for Release of Technical Information, Control Number: AFRL-PR-ED-VG-2001-134 Blanski, Rusty; Phillips, Shawn; Lee, Andre, "The Preparation and Properties of Polymer/Nanoparticle Blends Using POSS<sup>TM</sup>" (VuGraphs)

2001 International Symposium on Nanocomposites (Chicago, IL 25-27 June 2001)(Deadline: 24 June 2001)

(Statement A)

## Polymer/Nanoparticle Blends Using POSSTM The Preparation and Properties of



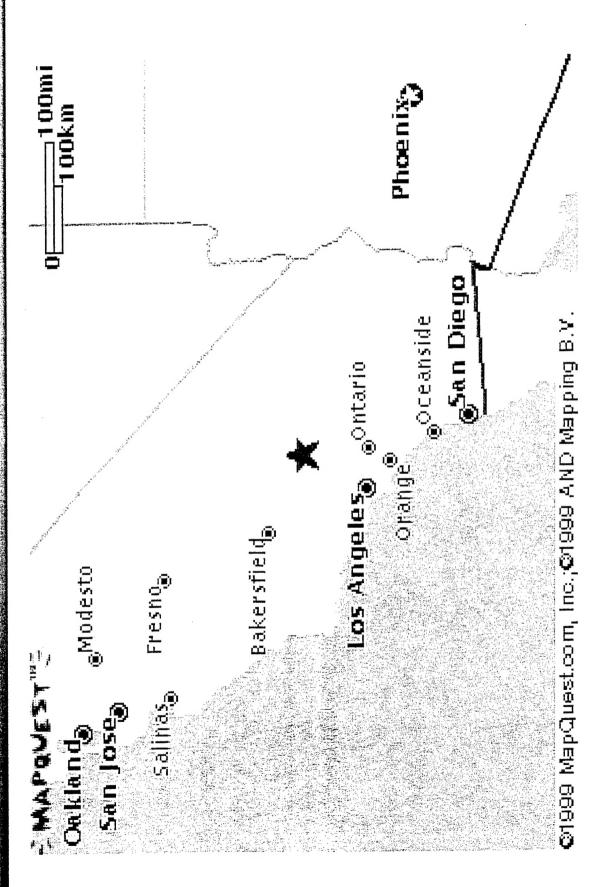
Rusty L. Blanski¹, Shawn H. Phillips¹, and Andre Lee²

<sup>1</sup>AFRL/PRSM, Air Force Research Laboratory, 10 E. Saturn Blvd, Bldg. 8451, Edwards AFB, CA 93524, <sup>2</sup>Department of Materials Science and Mechanics, Michigan State University, East Lansing, MI 48824



#### Air Force Research Laboratory Located ~ 100 miles from LA











candidates for blending: Polyethylene, Several polymers were looked into as Polystyrene, Polycarbonate, and Styrene-Butadiene Rubber က



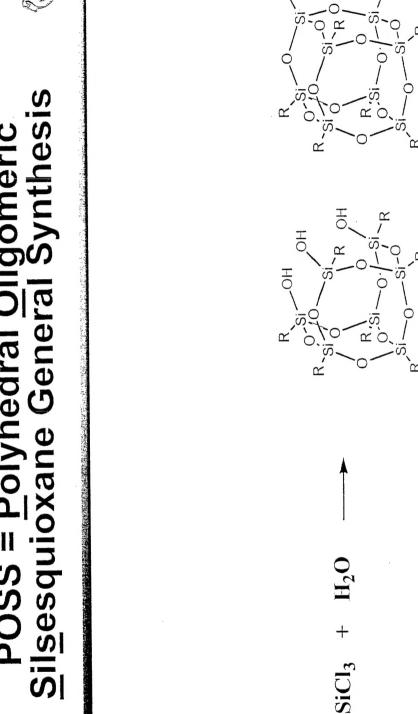
## Why Use Blendables?



- of the POSS molecule to give a polymer- Easier to tailor the organic side groups soluble species
- instead of copolymerization with reactive Simple blending techniques can be used POSS monomers
- without requiring expensive replacement Potential Drop-in molecular modifier of processing equipment

# POSS = Polyhedral Oligomeric Silsesquioxane General Synthesis





H<sub>2</sub>O

$$R = \frac{R}{S_{1}} = \frac{S_{1}}{S_{1}} = \frac{R}{S_{1}} = \frac{S_{1}}{S_{1}} = \frac{S_{1}$$

RSiCl<sub>3</sub> + H<sub>2</sub>O

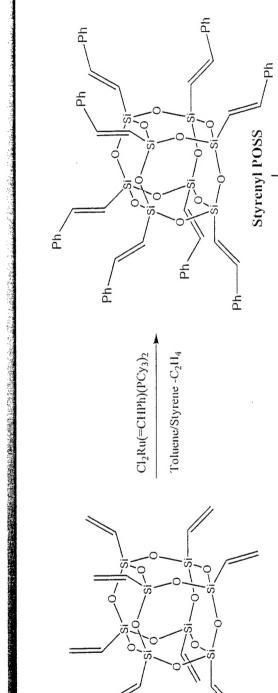
$$\begin{bmatrix}
R \\ Si \\ O \\ Si \\ O
\end{bmatrix}$$

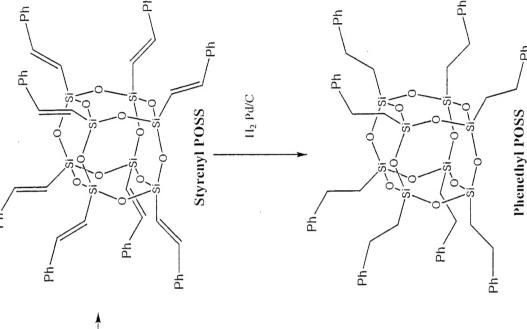
$$\begin{bmatrix}
R \\ Si \\ O
\end{bmatrix}$$

$$\begin{bmatrix}$$

# POSS = Polyhedral Oligomeric Silsesquioxane General Synthesis

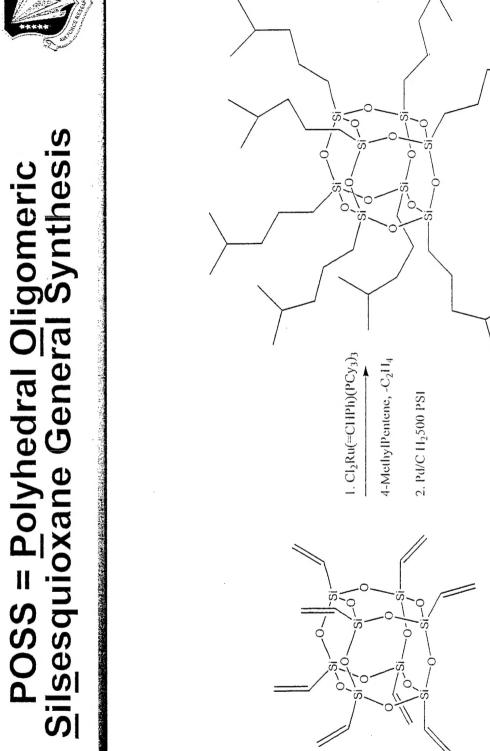


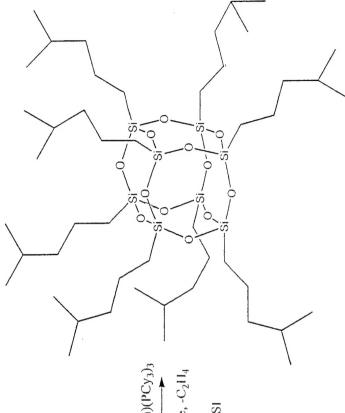












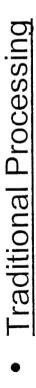
4-MP POSS





# Preparation of Polymer-POSS Blends



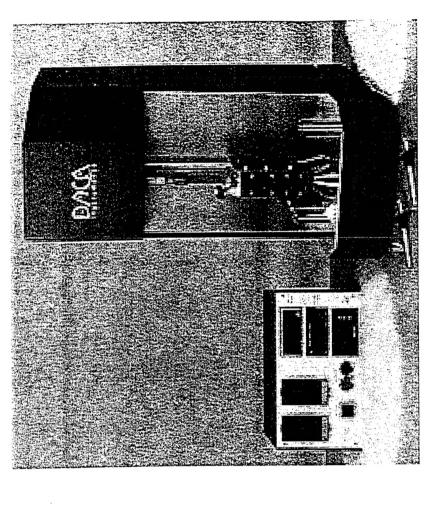


Place Polystyrene in Extruder

Add POSS

Blend 2-5 Minutes

Use a DACA for small scale (4 g)



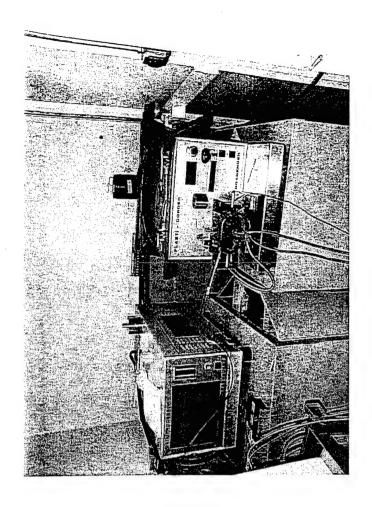




# Preparation of Polymer-POSS Blends



- Traditional Processing:
- Brabender Mixer
- Place Polystyrene in Mixer at temperature
- · Add POSS
- Blend 5-10 Minutes
- · Grind
- Press into disks/extrude/ injection mold

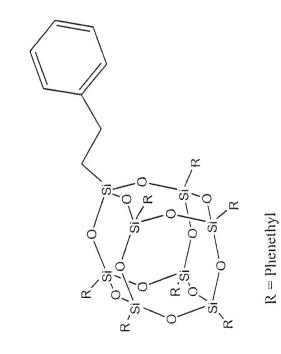


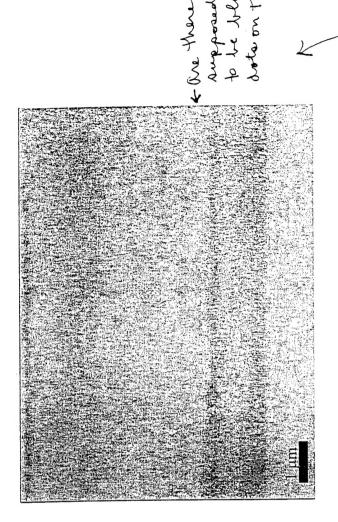


### **Miscibility of POSS**



50 wt % Phenethyl<sub>8</sub>T<sub>8</sub> in 2 million mol. wt. Polystyrene





- Demonstrated Complete Miscibility
- Grey domains represent miscible POSS/polystyrene
- Black dots are POSS crystallites (<100 POSS molecules)</li>
- 30% increase in surface hardness observed

## **POSS-Polymer Blends**



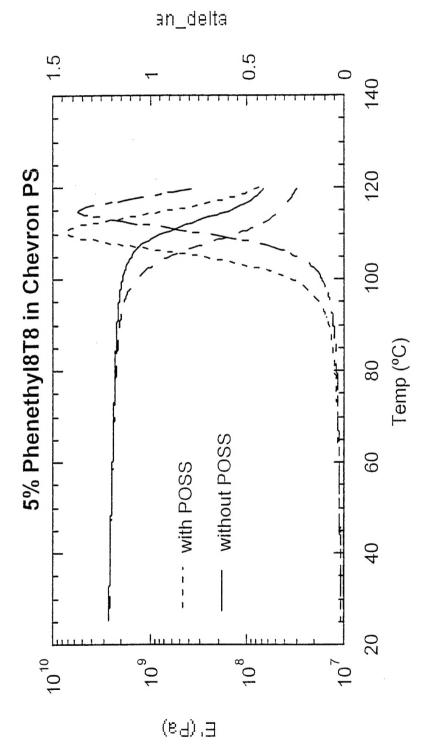
## Selected Data for POSS-Polymer Blends

Resin	POSS	Processing	Appearance
	compound	Temp. °C	
Polystyrene	Phenethyl <sub>8</sub> T <sub>8</sub>	177	Clear
BPA polycarbonate	Phenethyl $_8\mathrm{T}_8$	300	Clear
SB Rubber	Phenethyl <sub>8</sub> T <sub>8</sub>	100	Clear
HDPE	$Octyl_8T_8$	120	Cloudy
HDPE	$4\text{-MP}_8\mathrm{T}_8$ (2)	120	Cloudy



# Addition of POSS into PS





No Change in modulus at ambient Temperatures Small Change in T<sub>g</sub> observed



#### Conclusions



- POSS can be blended and dispersed into many polymers
- extremely important in determining the solubility of The organic side groups on the POSS molecule are the POSS in polymers
- The addition of the more soluble Styrenyl POSS into styrene leads to an increase in surface hardness without adversely affecting polymer properties
- POSS can be thought of as functionalized silicas with the side groups acting as solubility enhancers